

# ***CHEM 103: Chemistry in Context***

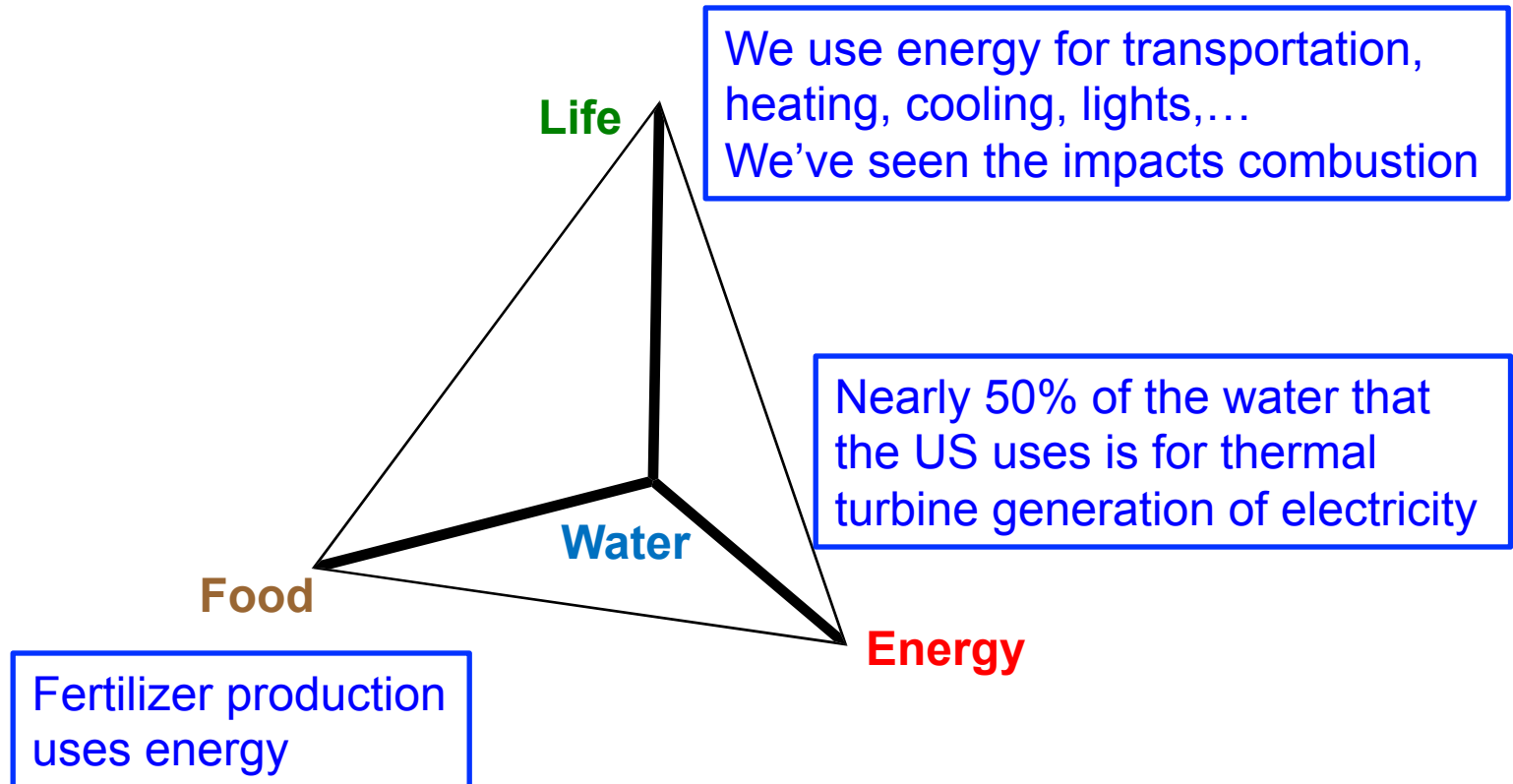
## *Unit 3 Energy Sources*

*Reading: CC Chapters 4, 7, 8,  
and G&R 4.1, 6, 5*

### *Unit 3.1 Introduction; Fossil Fuels*



# Interconnectedness of Issues: Energy



# Scale of Energy use



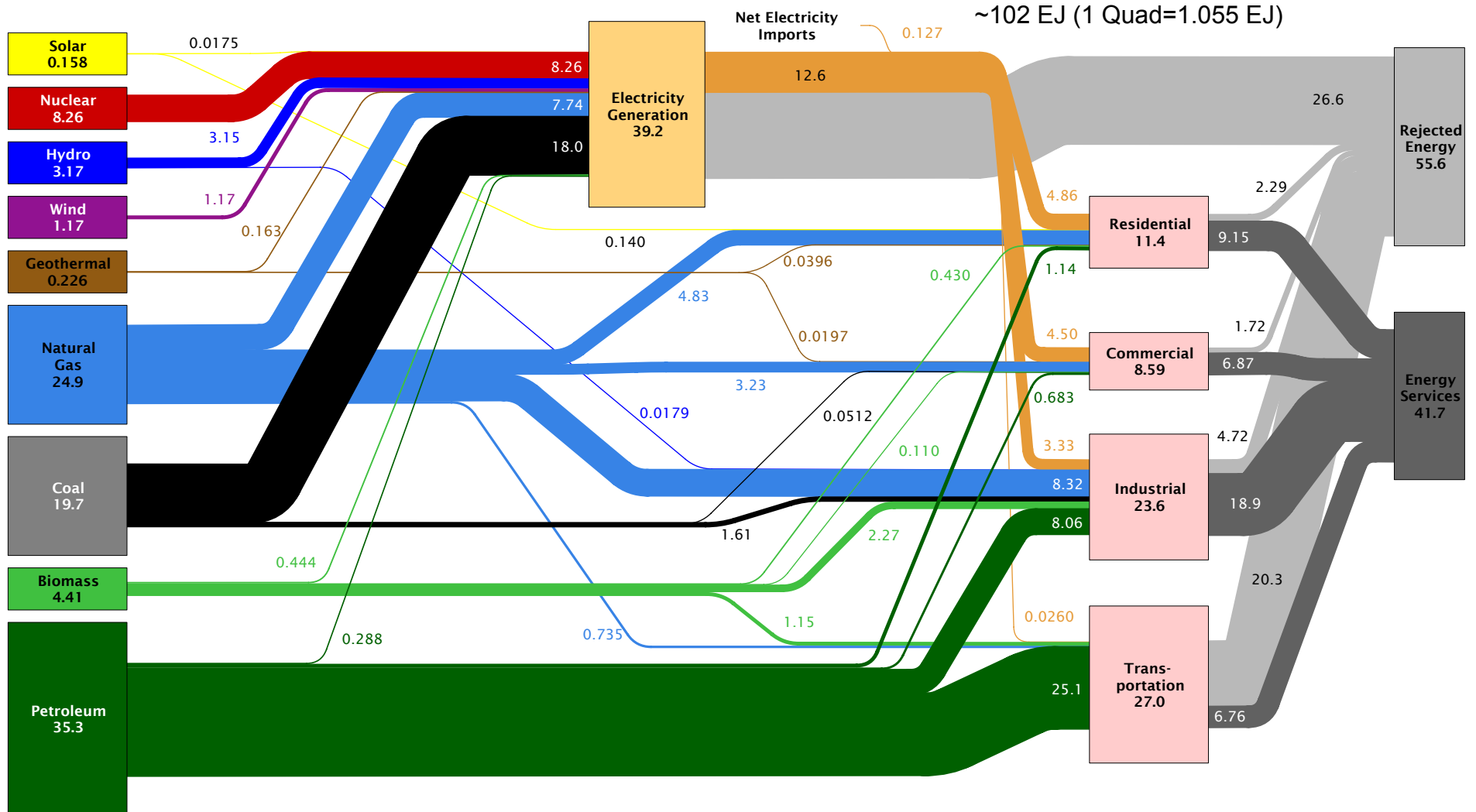
In 2011 the average world citizen used 77 GJ,  
the average US citizen used 315 GJ.

**TABLE 3.2**

ENERGY		ENERGY SOURCE OR USE
9 kJ	$10^4$	1.225V Alkaline AA Battery
310 kJ	$10^5$	Heating a quart (Liter) of water
1.2 MJ	$10^6$	Snickers bar
1.7 MJ	$10^6$	12V lead acid battery
10 MJ	$10^7$	Nutritional energy required by one person per day
140 MJ	$10^8$	Burning a log of wood
2 GJ	$10^9$	Energy in 15 gallons of gasoline
11.6 GJ	$10^{10}$	Energy to heat your house in Colorado in January
100 GJ	$10^{11}$	Annual US home (northern) heating and electric use
2.9 TJ	$10^{12}$	Energy in a coal train car full of coal
346 TJ	$10^{14}$	Energy in a coal train full of coal (120 cars)
9 PJ	$10^{16}$	Annual output from a 280 MW coal plant
32 PJ	$10^{16}$	Annual output from a 1 GW nuclear power plant
46 PJ	$10^{16}$	Energy for a small (pop. 144,000) US city
400 PJ	$10^{17}$	Energy for a large majority world city
4 EJ	$10^{18}$	Energy for a large developed world city
26 EJ	$10^{19}$	Caloric need to feed the world for a year
100 EJ	$10^{20}$	2011 US energy consumption
115 EJ	$10^{20}$	2011 China energy consumption
540 EJ	$10^{21}$	2011 World energy consumption
1 ZJ	$10^{21}$	Expected world energy demand in 2025

# Energy Flow

Estimated U.S. Energy Use in 2011: ~97.3 Quads



Source: LLNL 2012. Data is based on DOE/EIA-0384(2011), October, 2012. If this information or a reproduction of it is used, credit must be given to the Lawrence Livermore National Laboratory and the Department of Energy, under whose auspices the work was performed. Distributed electricity represents only retail electricity sales and does not include self-generation. EIA reports flows for non-thermal resources (i.e., hydro, wind and solar) in BTU-equivalent values by assuming a typical fossil fuel plant "heat rate." The efficiency of electricity production is calculated as the total retail electricity delivered divided by the primary energy input into electricity generation. End use efficiency is estimated as 80% for the residential, commercial and industrial sectors, and as 25% for the transportation sector. Totals may not equal sum of components due to independent rounding. LLNL-MI-410527

# Energy Transformation

## Types of energy:

**Potential:** stored energy/  
energy of position

**Kinetic:** energy of motion

## Forms of energy:

Mechanical

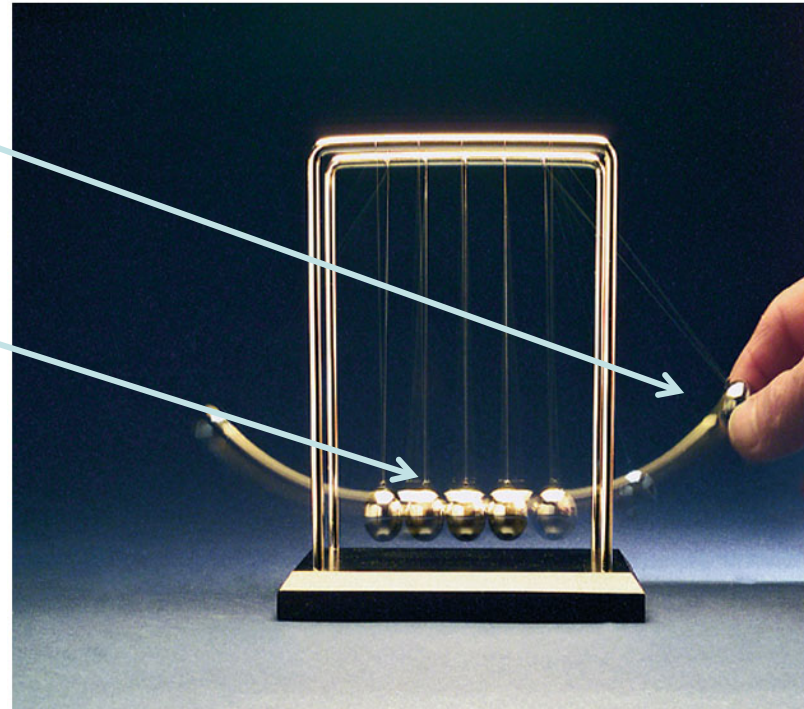
Radiant (light)

Electrical

Chemical

Nuclear

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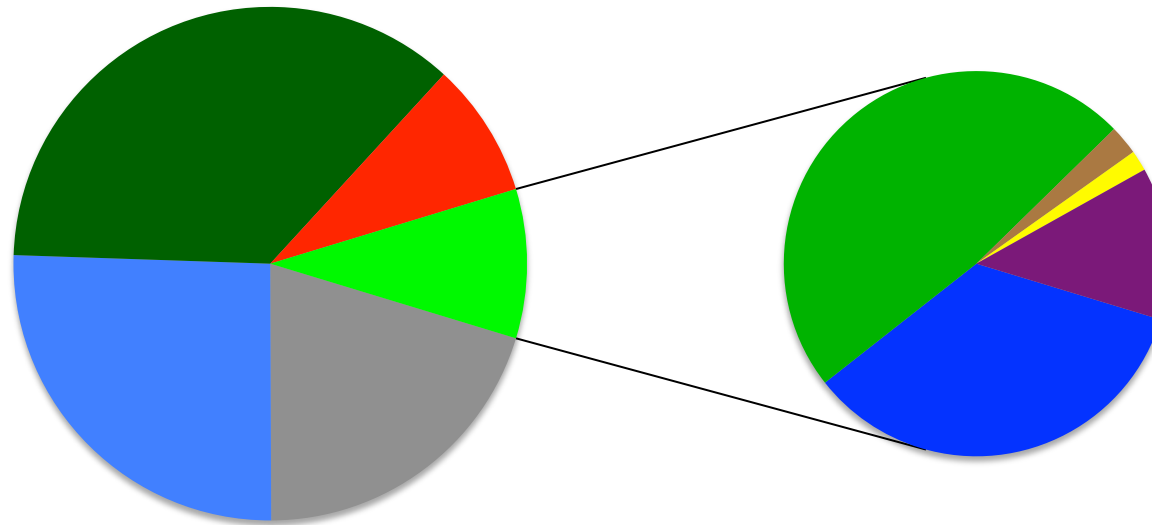
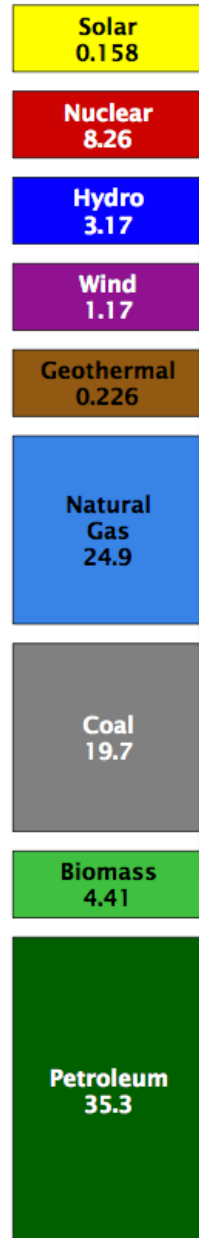


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**1<sup>st</sup> Law of thermodynamics:** Energy is neither created nor destroyed (but we can & do transform it)

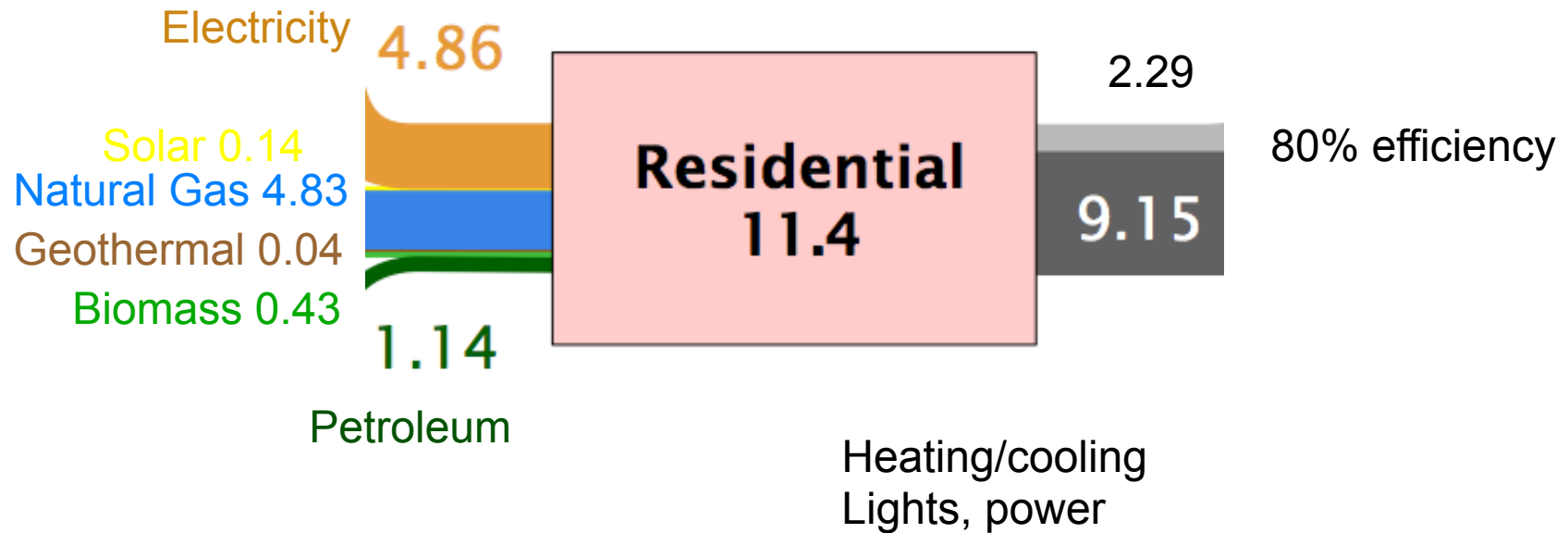
**2<sup>nd</sup> Law of thermodynamics:** Entropy always increases (limits efficiency of heat-driven processes)

# Energy Sources

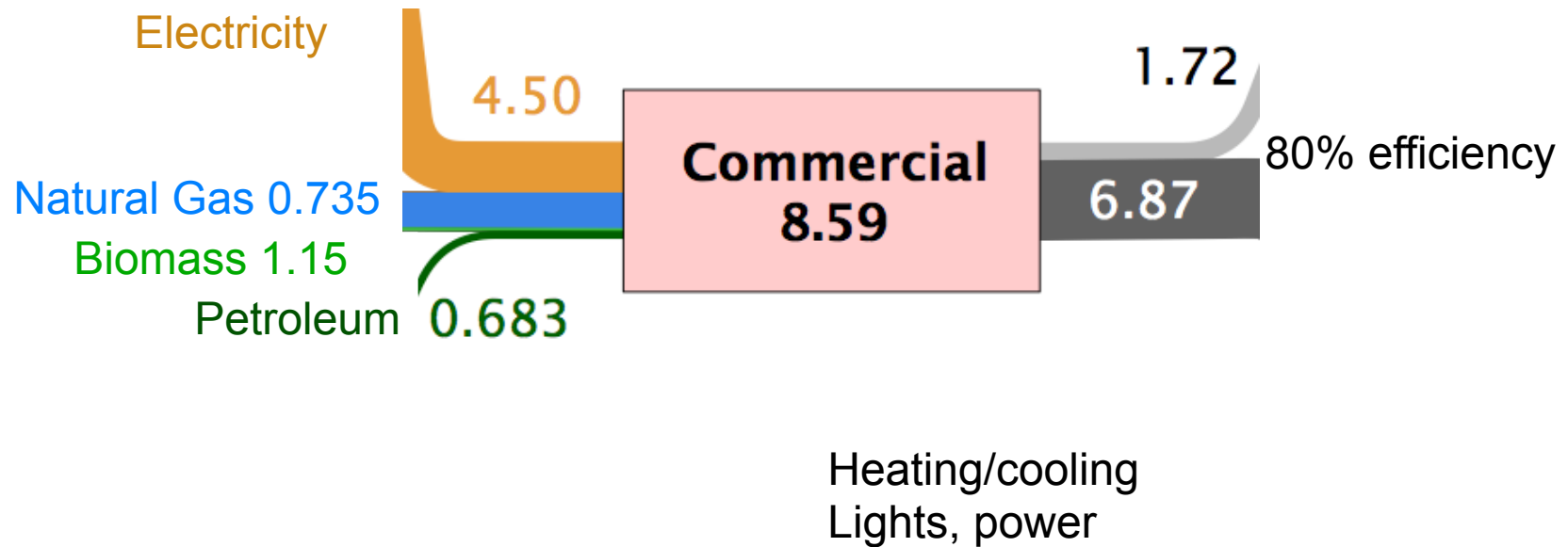


- Coal
- Natural Gas
- Petroleum
- Nuclear
- Hydroelectric
- Biomass
- Geothermal
- Solar
- Wind

## Energy Uses: Residential

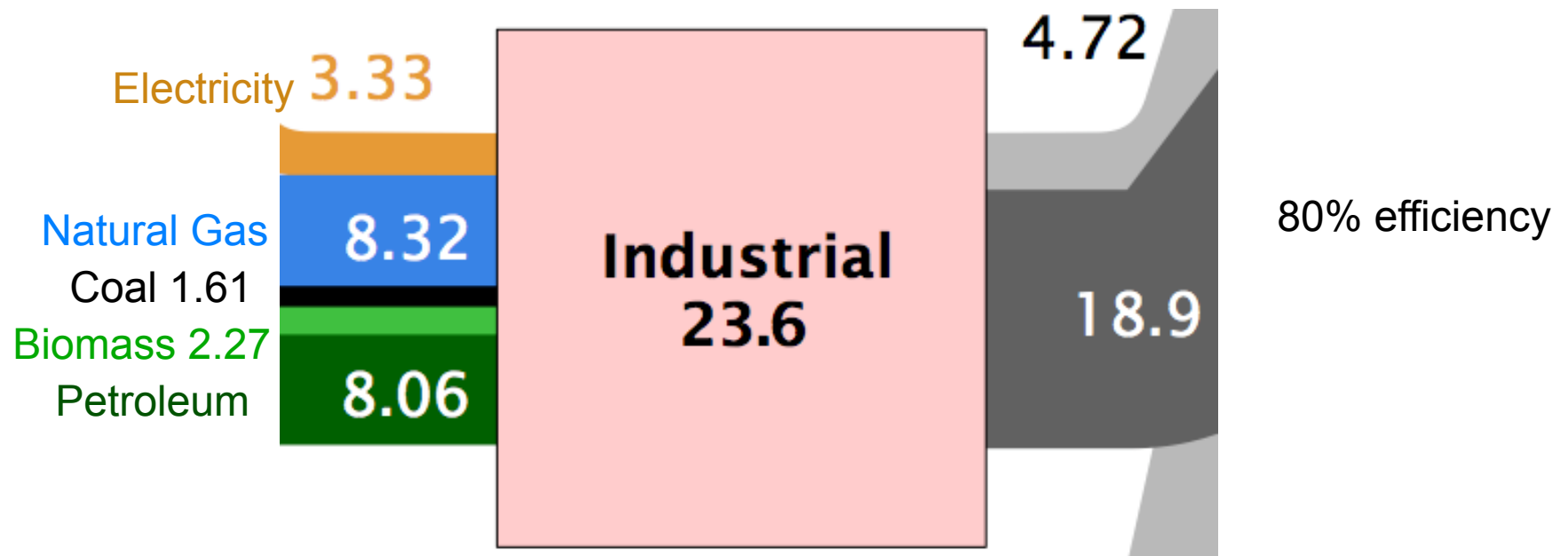


## Energy Uses: Commercial



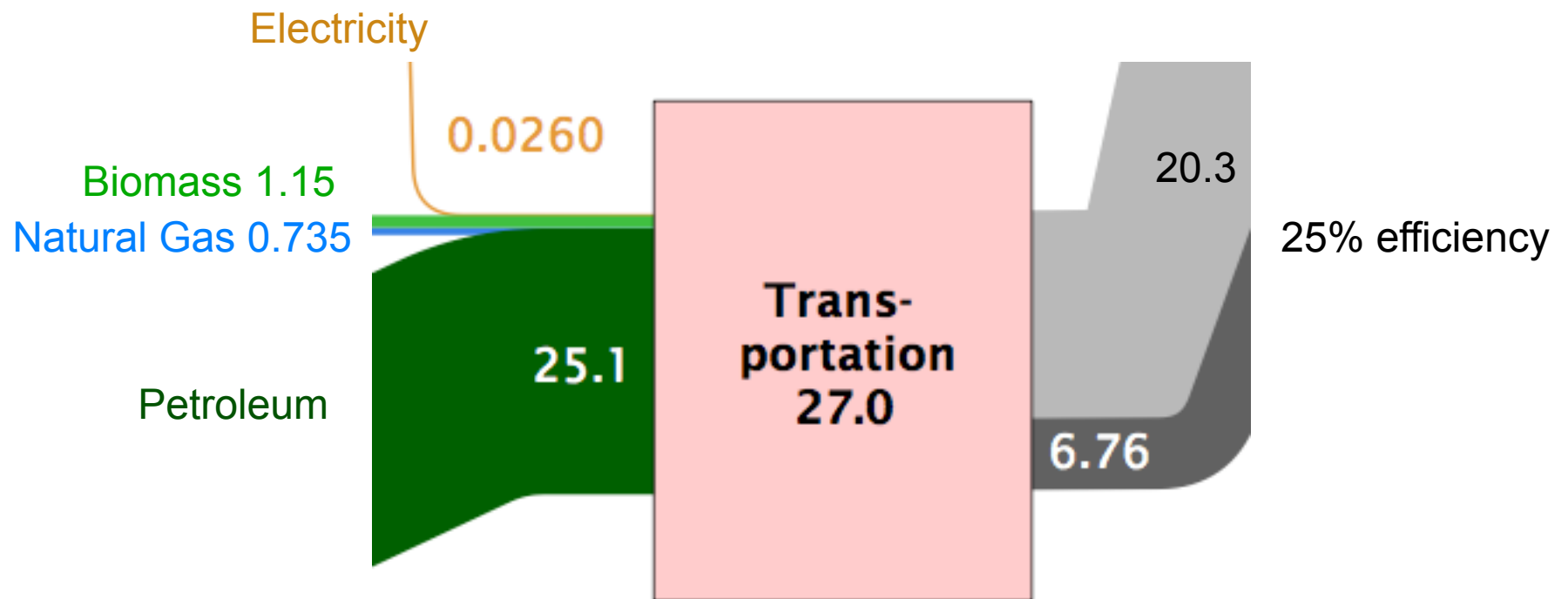


## Energy Uses: Industrial

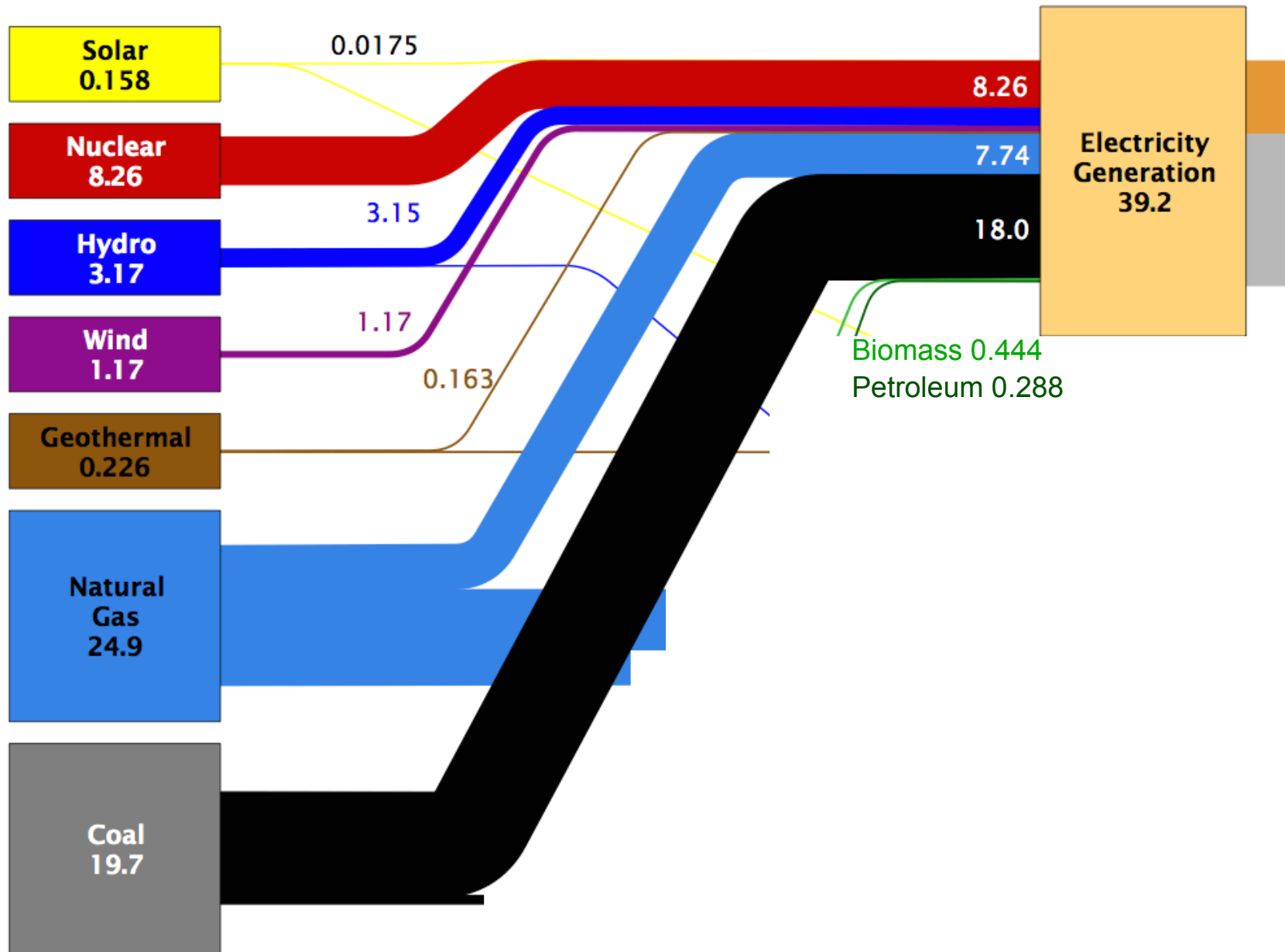


Manufacturing (steel, plastics, electronics,...)

## Energy Uses: Transportation



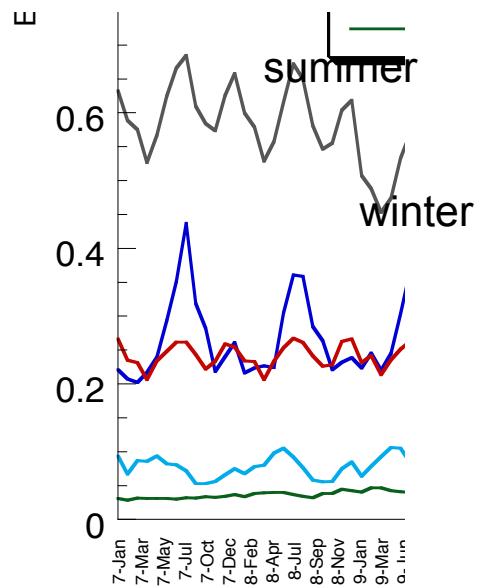
# Electricity sources



# Electricity efficiency



# Electricity Seasonal uses & sources



# Power Plant

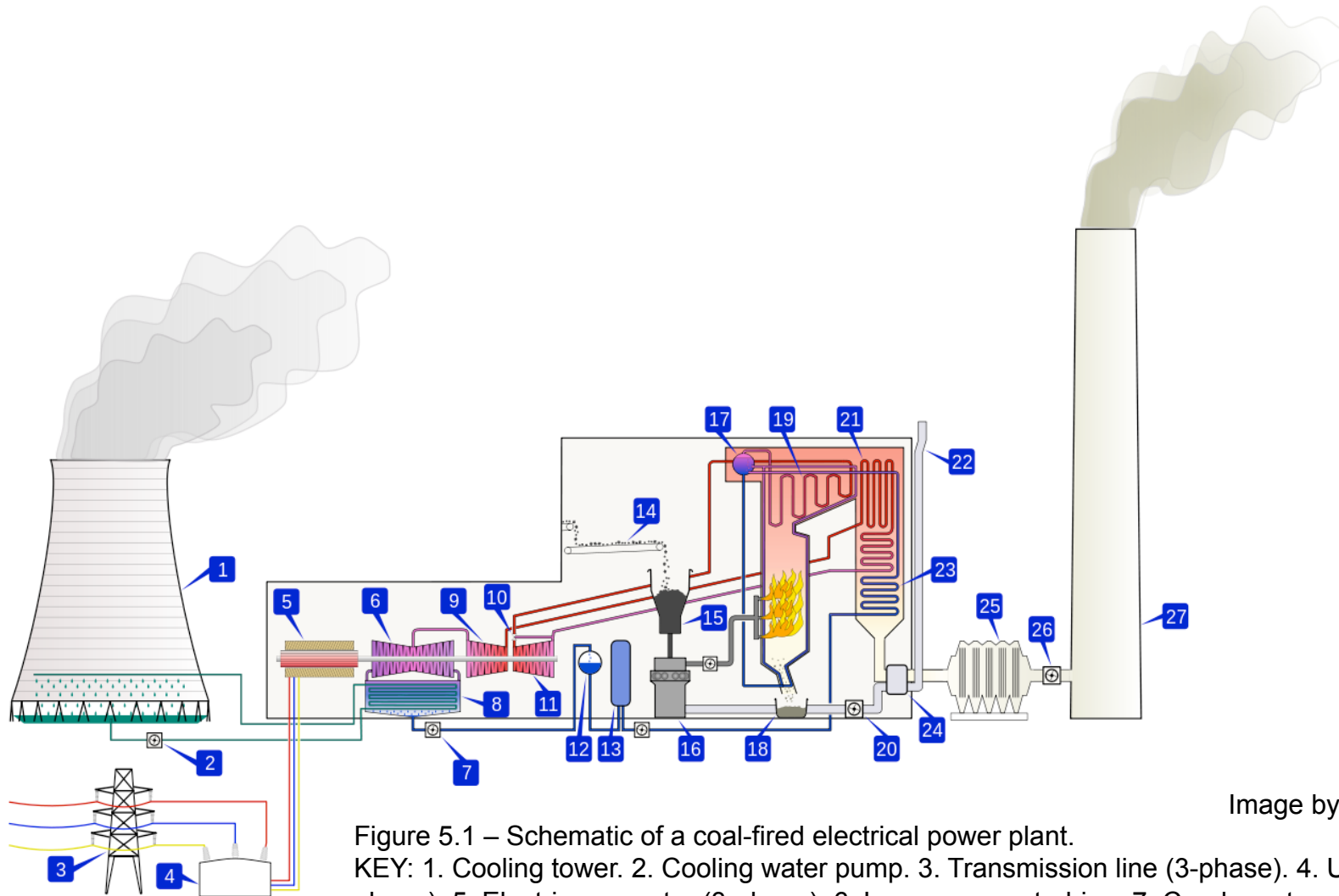


Image by BillC/CC BY-SA 3.0

Figure 5.1 – Schematic of a coal-fired electrical power plant.

KEY: 1. Cooling tower. 2. Cooling water pump. 3. Transmission line (3-phase). 4. Unit transformer (3-phase). 5. Electric generator (3-phase). 6. Low pressure turbine. 7. Condensate extraction pump. 8. Condenser. 9. Intermediate pressure turbine. 10. Steam governor valve. 11. High pressure turbine. 12. Deaerator. 13. Feed heater. 14. Coal conveyor. 15. Coal hopper. 16. Pulverised fuel mill. 17. Boiler drum. 18. Ash hopper. 19. Superheater. 20. Forced draught fan. 21. Reheater. 22. Air intake. 23. Economiser. 24. Air preheater. 25. Precipitator. 26. Induced draught fan. 27. Chimney Stack.

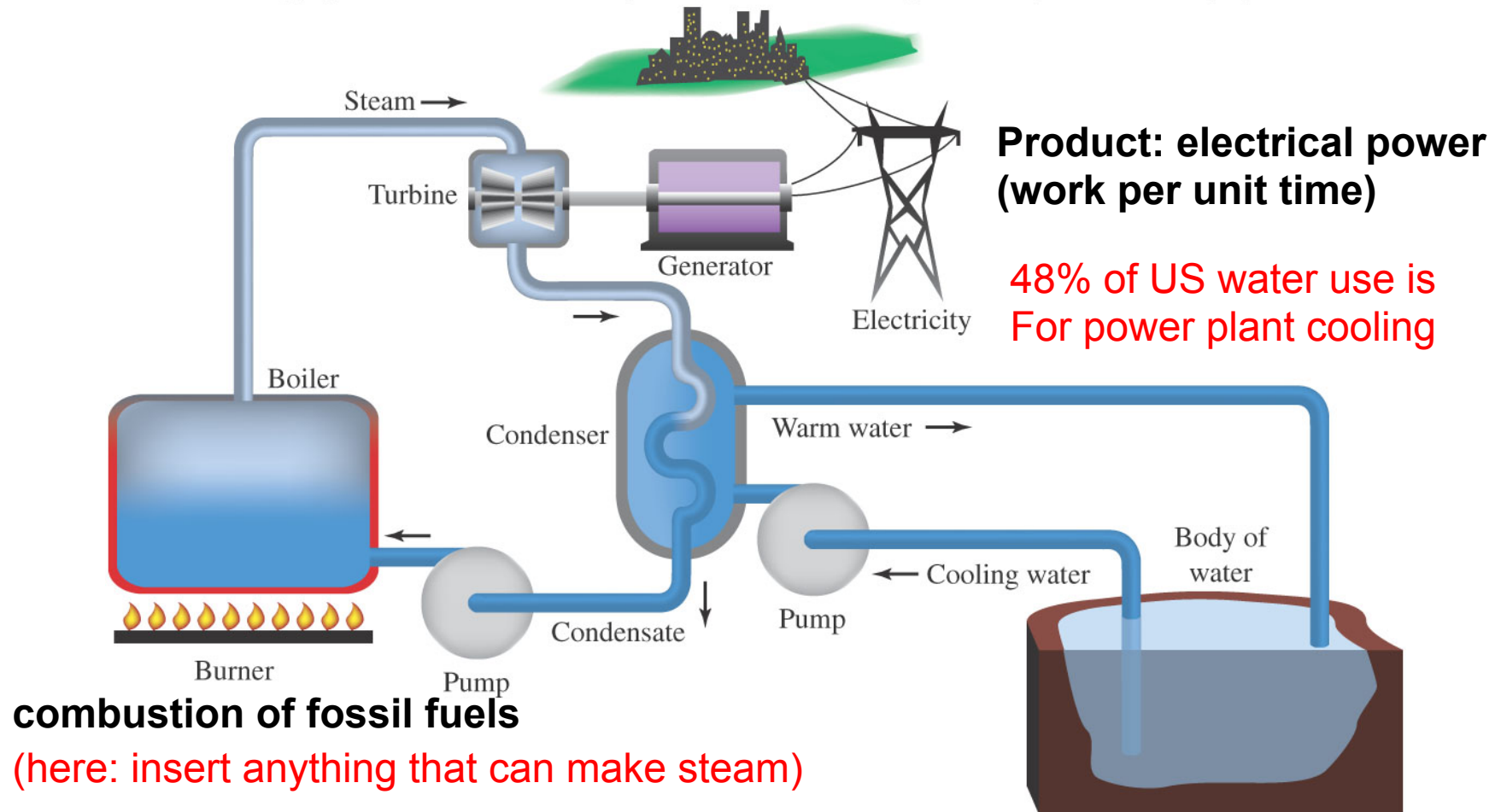
# Rawhide Power Plant



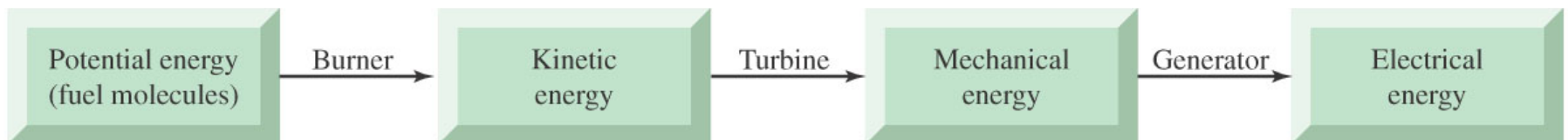
Image ©2012 DigitalGlobe, GeoEye, USDA Farm Service Agency, Map data ©2013 Google

**FIGURE 5.2** – Google Maps satellite image of the Rawhide Power Plant of the Platte River Power Authority

# Power Plant and Energy Transformation

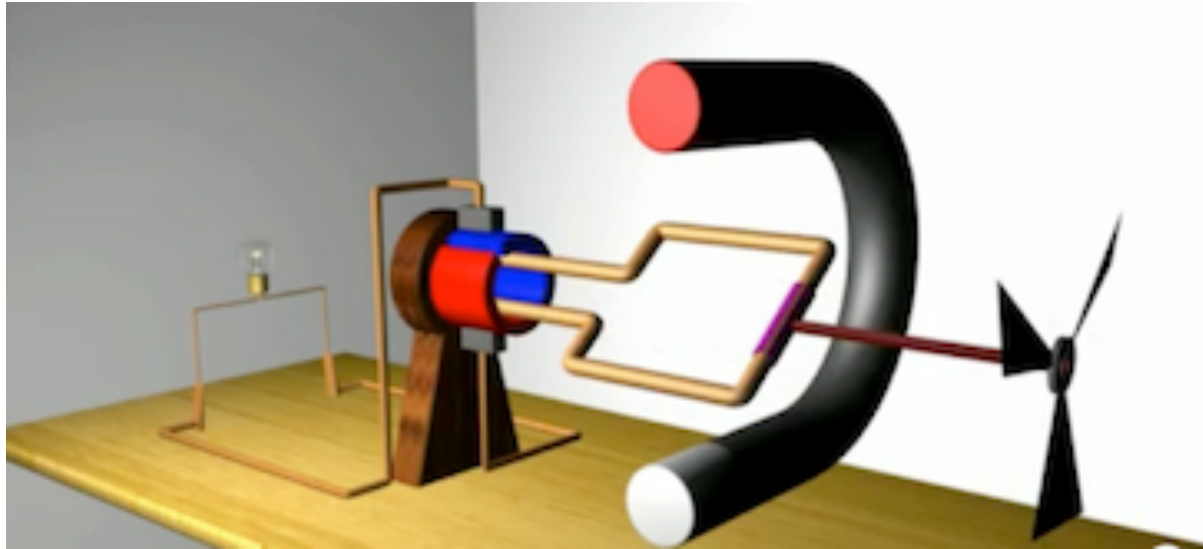


Energy transformation process:





# Electric Generator



Nuclear

Fossil Fuel (coal, natural gas)

Solar thermal

} motion of steam

Hydroelectric

Wind

Tides

} motion of water/air

[http://www.youtube.com/watch?v=d\\_aTC0iKO68](http://www.youtube.com/watch?v=d_aTC0iKO68)

# Carnot Efficiency

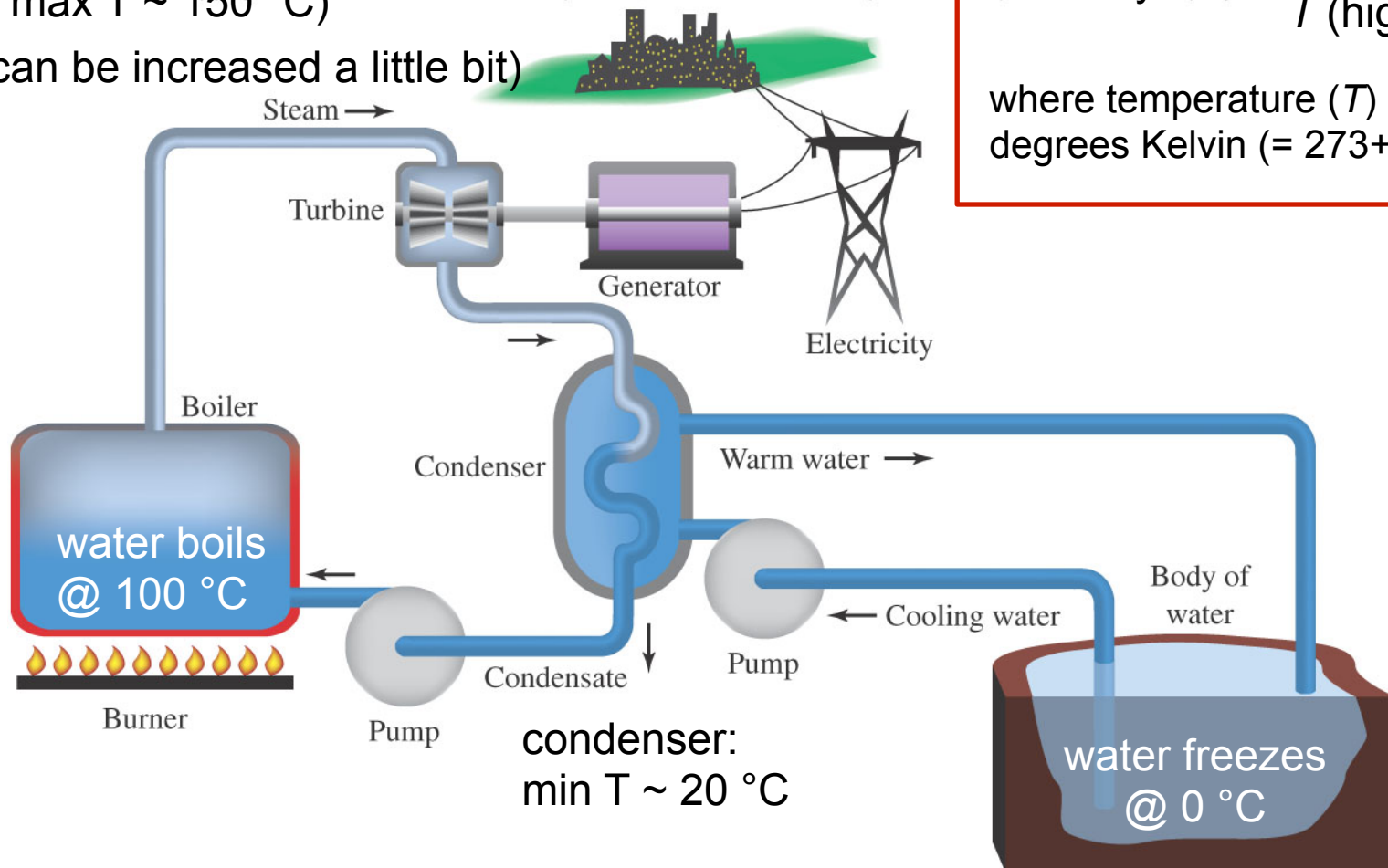
For any heat engine, the efficiency is related to the operating temperatures:

boiler technology:  
max  $T \sim 150\text{ }^{\circ}\text{C}$

(this can be increased a little bit)

$$\text{efficiency} = 1 - \frac{T(\text{low})}{T(\text{high})}$$

where temperature ( $T$ ) in  
degrees Kelvin ( $= 273 + ^{\circ}\text{C}$ )



# Energy and Efficiency: a Worked Example

*How much natural gas is burnt to keep your house warm?*

Givens:

1. It takes  $3.5 \times 10^7$  kJ to heat a northern US house in January
2. Electric heaters are 98% efficient at converting electrical energy into heat
3. Combustion of 1 g of methane releases 56 kJ of energy (calculated)

Power plants have limited efficiencies as well:

**2<sup>nd</sup> Law of thermodynamics:** all energy conversions suffer from inefficiencies

**Table 4.1**

**Typical Efficiencies in Power Production**

Maximum theoretical efficiency	55–60%	30% more realistic
Efficiency of boiler	90%	
Mechanical efficiency of turbine	75%	
Efficiency of electrical generator	95%	
Efficiency of power transmission	90%	
	(entropy limited)	

$$\text{Overall efficiency} = 0.6 \times 0.9 \times 0.75 \times 0.95 \times 0.9 \times 0.98 = 0.34$$

Heat used x efficiency = heat needed (i.e. the heat produced at the power plant)

$$\text{Heat used} \times 0.34 = 3.5 \times 10^7 \text{ kJ}$$

$$\rightarrow \text{Heat needed} = 1.0 \times 10^8 \text{ kJ}$$

$$1.0 \times 10^8 \text{ kJ} \times \frac{1 \text{ g CH}_4}{56 \text{ kJ}} = 1.8 \times 10^6 \text{ g CH}_4$$

**2 metric tons!**