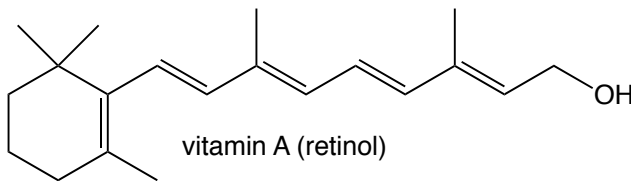


# You are what you eat

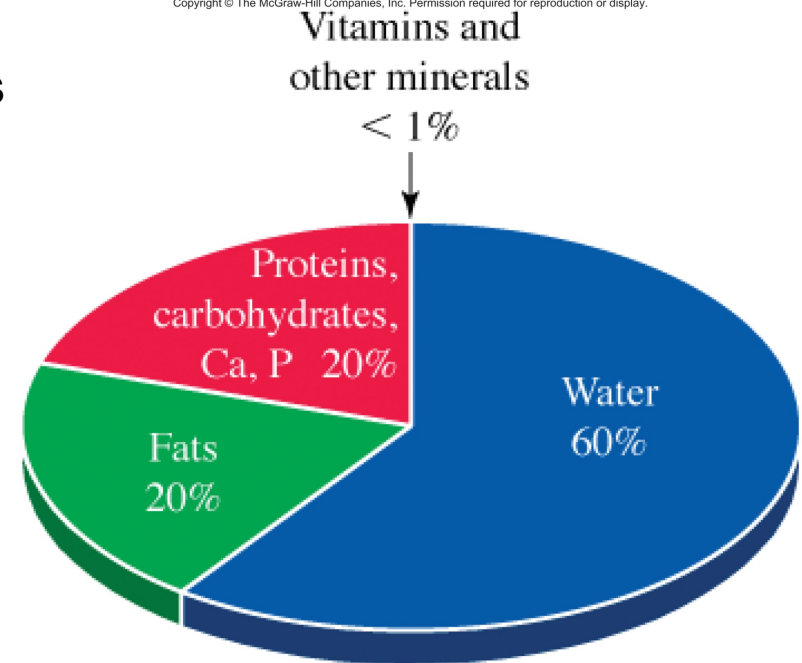
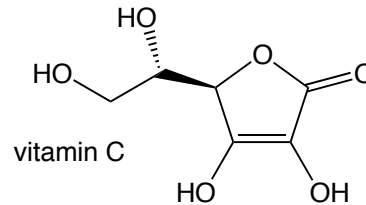
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Macronutrients: fats, proteins, carbohydrates  
Essential amino acids

Vitamins and minerals  
Hydrophobic



Hydrophilic



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Table 11.1 Percent Water, Fat, Carbohydrates, and Protein in Selected Foods				
Food	Water	Fat	Carbohydrates	Protein
white bread	37	4	48	8
2% milk	89	2	5	3
chocolate chip cookies	3	23	69	4
peanut butter	1	50	19	25
sirloin steak	57	15	0	28
tuna fish	63	2	0	30
black beans (cooked)	66	<1	23	9

Source: U.S. Department of Agriculture, Agricultural Research Service, *Home and Garden Bulletin 72*, 2002.

## Nutritional periodic Table

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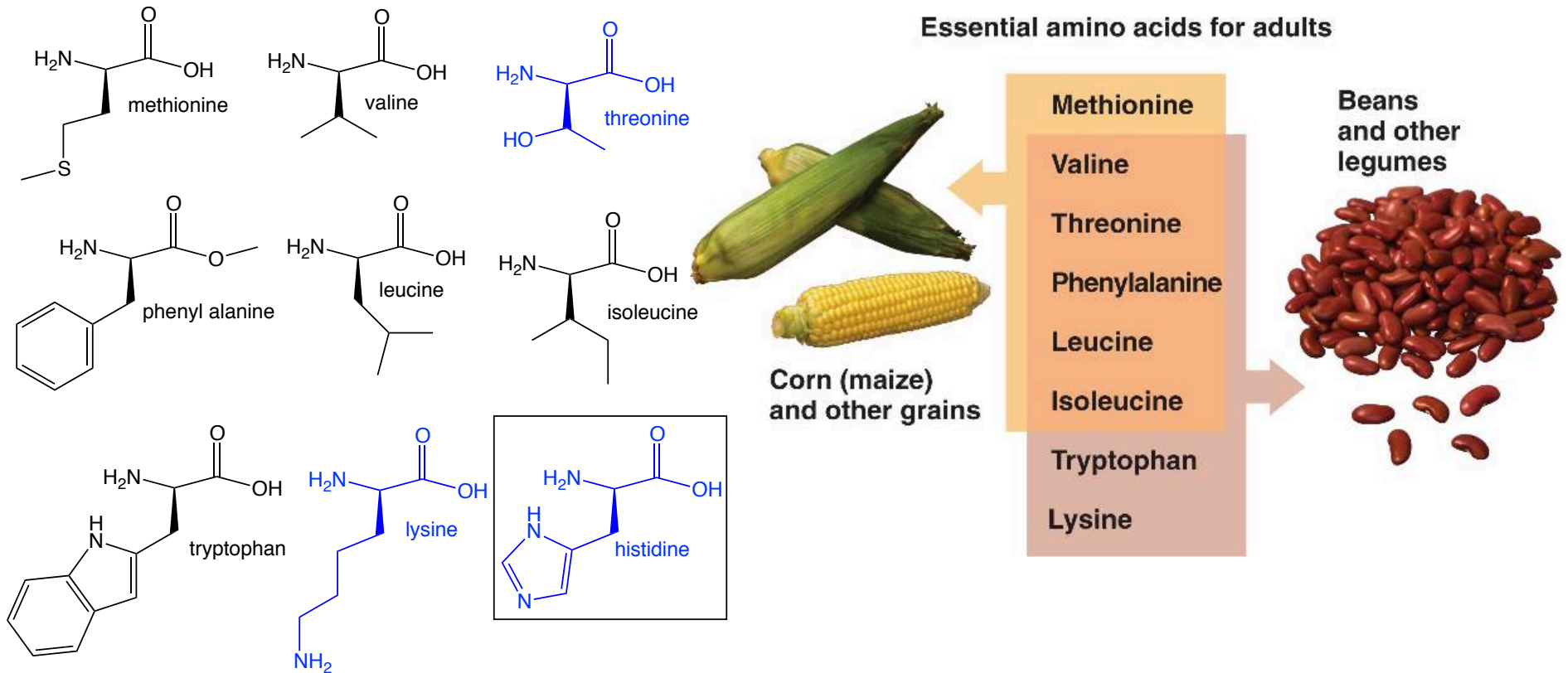
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## Macrominerals

## Microminerals

## Trace minerals

# Essential amino acids



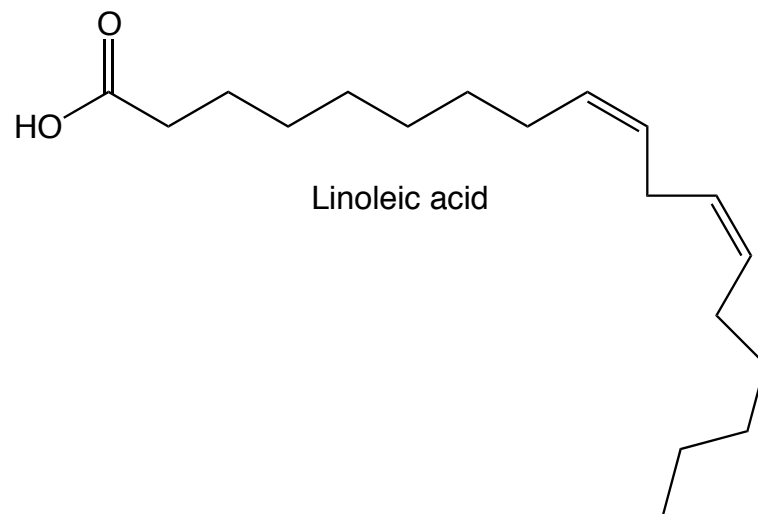
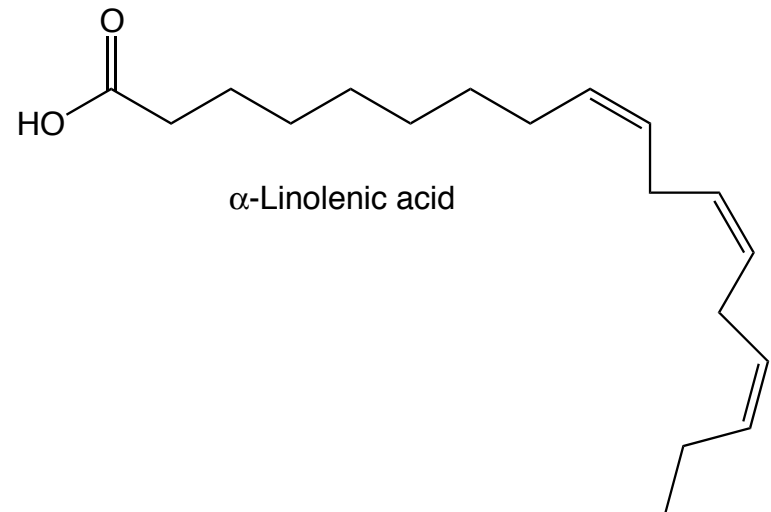
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**Table 11.5**

**The Essential Amino Acids**

histidine	lysine	threonine
isoleucine	methionine	tryptophan
leucine	phenylalanine	valine

# Essential fatty acids



# Caloric “Need”

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Table 11.7	Estimated Calorie Requirements (United States)		
ACTIVITY LEVEL			
Age (yr)	Sedentary*	Moderately Active <sup>†</sup>	Active <sup>‡</sup>
<b>Females</b>			
14–18	1800	2000	2400
19–30	2000	2000–2200	2400
31–50	1800	2000	2200
51+	1600	1800	2000–2200
<b>Males</b>			
14–18	2200	2400–2800	2800–3200
19–30	2400	2600–2800	3000
31–50	2200	2400–2600	2800–3000
51+	2000	2200–2400	2400–2800

\**Sedentary* means a lifestyle that includes only the light physical activity associated with typical day-to-day life.

† *Moderately active* means a lifestyle that includes physical activity equivalent to walking about 1–3 miles per day at 3–4 miles per hour, in addition to the light physical activity associated with typical day-to-day life.

‡ *Active* means a lifestyle that includes physical activity equivalent to walking more than 3 miles per day at 3–4 miles per hour, in addition to the light physical activity associated with typical day-to-day life.

Source: *Dietary Guidelines for Americans*, 2005, USDA.

# Energy use

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**Table 11.8**

**Energy Expenditure for Common Physical Activities\***

Moderate Physical Activity	Cal/hr	Vigorous Physical Activity	Cal/hr
hiking	370	jogging (5 mph)	590
light gardening/yard work	330	heavy yard work (chopping wood)	440
dancing	330	swimming (freestyle laps)	510
golf (walking, carrying clubs)	330	aerobics	480
bicycling (<10 mph)	290	bicycling (>10 mph)	590
walking (3.5 mph)	280	walking (4.5 mph)	460
weight lifting (light workout)	220	weightlifting (vigorous workout)	440
stretching	180	basketball (vigorous)	440

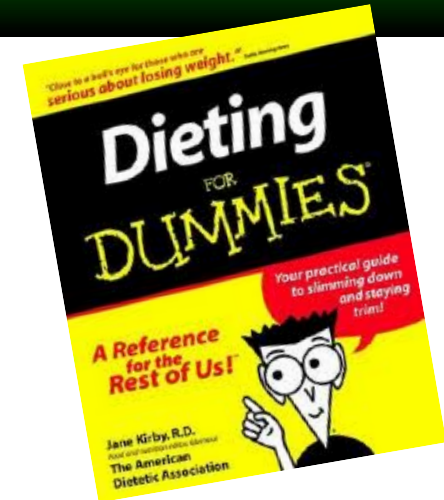
\* Values include both resting metabolic rate and activity expenditure for a 70-kg (154-pound) person. Calories burned per hour are higher for persons heavier than 154 pounds and lower for persons who weigh less.

## Feeling the heat

How long would you have to walk to “burn off” a snickers bar?  
(271 Calories per snickers; 280 Calories per hour for walking)

$$271 \text{ Calories} \times \frac{1 \text{ hr}}{280 \text{ Calories}} = 0.97 \text{ hr}$$

# Feeling the Heat

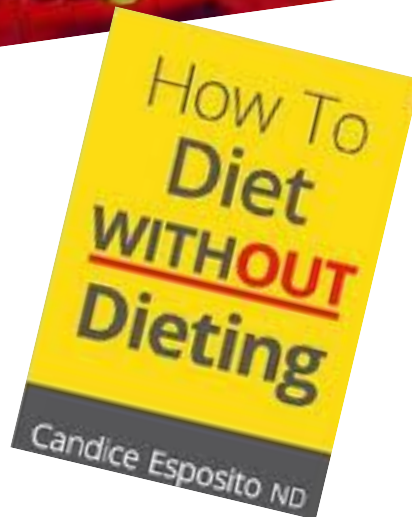


The average human needs about 2000 Cal per day.  
**Times 4 that's ~8000 kJ per day.**

**8 Snickers bars! 12 12 oz. Cokes!**

**Want a more balanced diet?**

**How about 6 Snickers bars and 3 Cokes?**



**9 Cal/g → 3500 Calories/lb**

## Feeling the heat

How long would you have to jog to lose a pound of fat?  
(3500 Calories per pound; 590 Calories per hour for jogging)

$$3500 \text{ Calories} \times \frac{1 \text{ hr}}{590 \text{ Calories}} = 5.9 \text{ hr}$$

# Feeling the Heat Globally



## World Population

**7 billion**

As of October 31, 2011



**China 1.3 billion (19.2%)**

India 1.2 billion (17%)

European Union 500 million (7.1%)

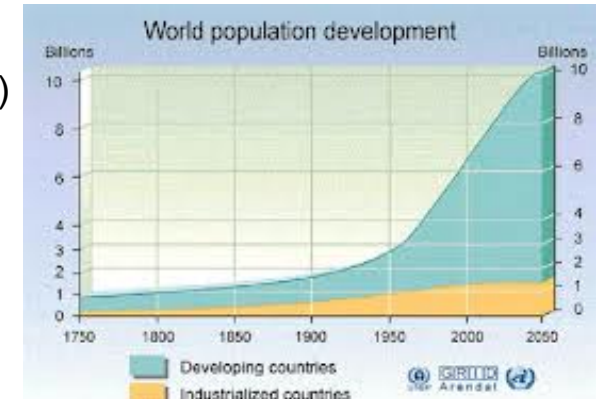
**USA 315 million (4.5%)**

Indonesia 238 million (3.3%)

Brazil 197 million (2.8%)

Pakistan 181 million (2.6%)

Nigeria 170 million (2.4%)



## Feeling the Heat Globally

### Feeding the World

**7 billion ( $7 \times 10^9$ ) people**

As of October 31, 2011

**8000 kJ per person per day**

**3 million ( $3 \times 10^6$ ) kJ per person per year**

$$7 \text{ billion} \times 3 \text{ million} = 21 \times 10^{15} \text{ kJ} = 21 \times 10^{18} \text{ J}$$

$$1 \text{ Exajoule (EJ)} = 10^{18} \text{ J}$$

**21 EJ to feed the world**

$$21 \times 10^{18} \text{ J} \times 1 \text{ tank of gas} / 2.1 \times 10^9 \text{ J} = 10 \text{ billion}$$

(US passenger fleet = 250 million; fill up 50 times/year = 12.5 billion)

$$21 \times 10^{18} \text{ J} \times 1 \text{ Snickers bar} / 1.0 \times 10^6 \text{ J} = 21 \text{ trillion (2 billion sold in 2011)}$$

## Feeling the Heat Globally

### Energy Use in 2011

**7 billion ( $7 \times 10^9$ ) people**

As of October 31, 2011

**21 EJ to feed the world**

**World 540 EJ – 100% (with 1.3 billion people) 77 GJ/person**

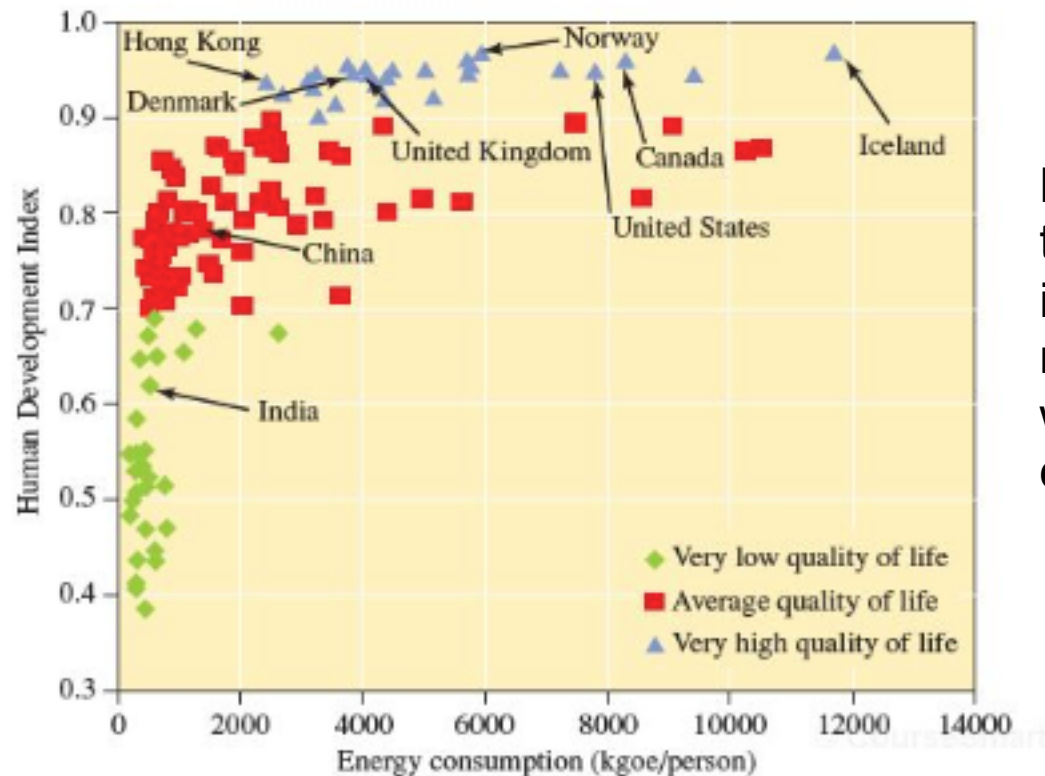
**China 115 EJ – 21.3% (with 1.3 billion people) 88 GJ/person**

**USA 100 EJ – 18.5% (with 315 million people) 317 GJ/person**

**EU 74 EJ – 13.8% (with 500 million people) 148 GJ/person**

**Former SU 45 EJ – 8.3% (with 285 million people) 158 GJ/person**

# Feeling the Heat Globally



**Figure 4.26**

The relationship between Human Development Index and energy consumption. Energy consumption is reported in kilograms of oil equivalent (kgoe) per person. 1 kgoe =  $4.4 \times 10^{10}$  J or the approximate energy released in burning a kilogram of oil.

Source: United Nations Human Development Report 2007/2008.

1 kgoe –  $4.4 \times 10^7$  J (not  $10^{10}$  as the caption says)

1000 kgoe/person = 44 GJ/person

Based on countries shown on the graph such as Hong Kong, it is possible to substantially reduce energy consumption without necessarily decreasing quality of life.

# Reactive Nitrogen

Dinitrogen gas is 78% of the atmosphere, but it's inert (i.e. useless) to the vast majority of living things...reactive forms are more useful but also potentially more dangerous:

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**Table 6.3**

**Some Reactive Forms of Nitrogen**

Name	Chemical Formula
nitrogen monoxide	NO
nitrogen dioxide	NO <sub>2</sub>
nitrous oxide → greenhouse gas	N <sub>2</sub> O
nitrate ion	NO <sub>3</sub> <sup>-</sup>
nitrite ion	NO <sub>2</sub> <sup>-</sup>
nitric acid	HNO <sub>3</sub>
ammonia → fertilizer	NH <sub>3</sub>
ammonium ion → in aerosols	NH <sub>4</sub> <sup>+</sup>

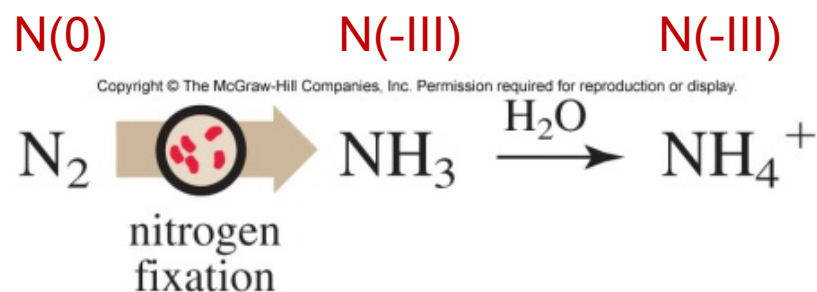
*pollutants* (pointing to NO, NO<sub>2</sub>, N<sub>2</sub>O)

*Plants can use these forms* (pointing to NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, NH<sub>3</sub>, NH<sub>4</sub><sup>+</sup>)

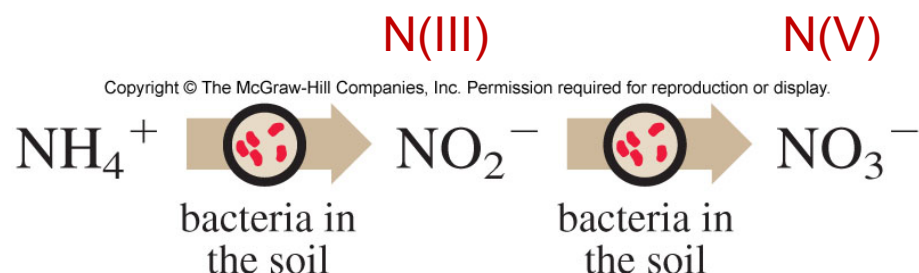
Note: All are naturally occurring.

# Redox in the Nitrogen Cycle

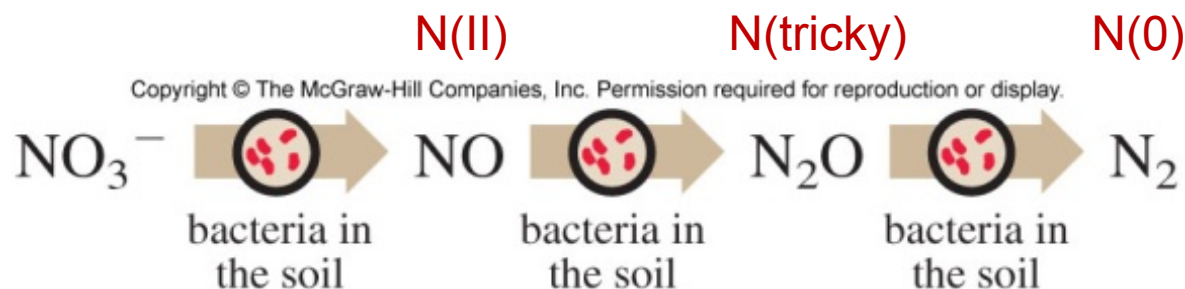
- 1) Nitrogen fixation by soil bacteria



- 2) Nitrification

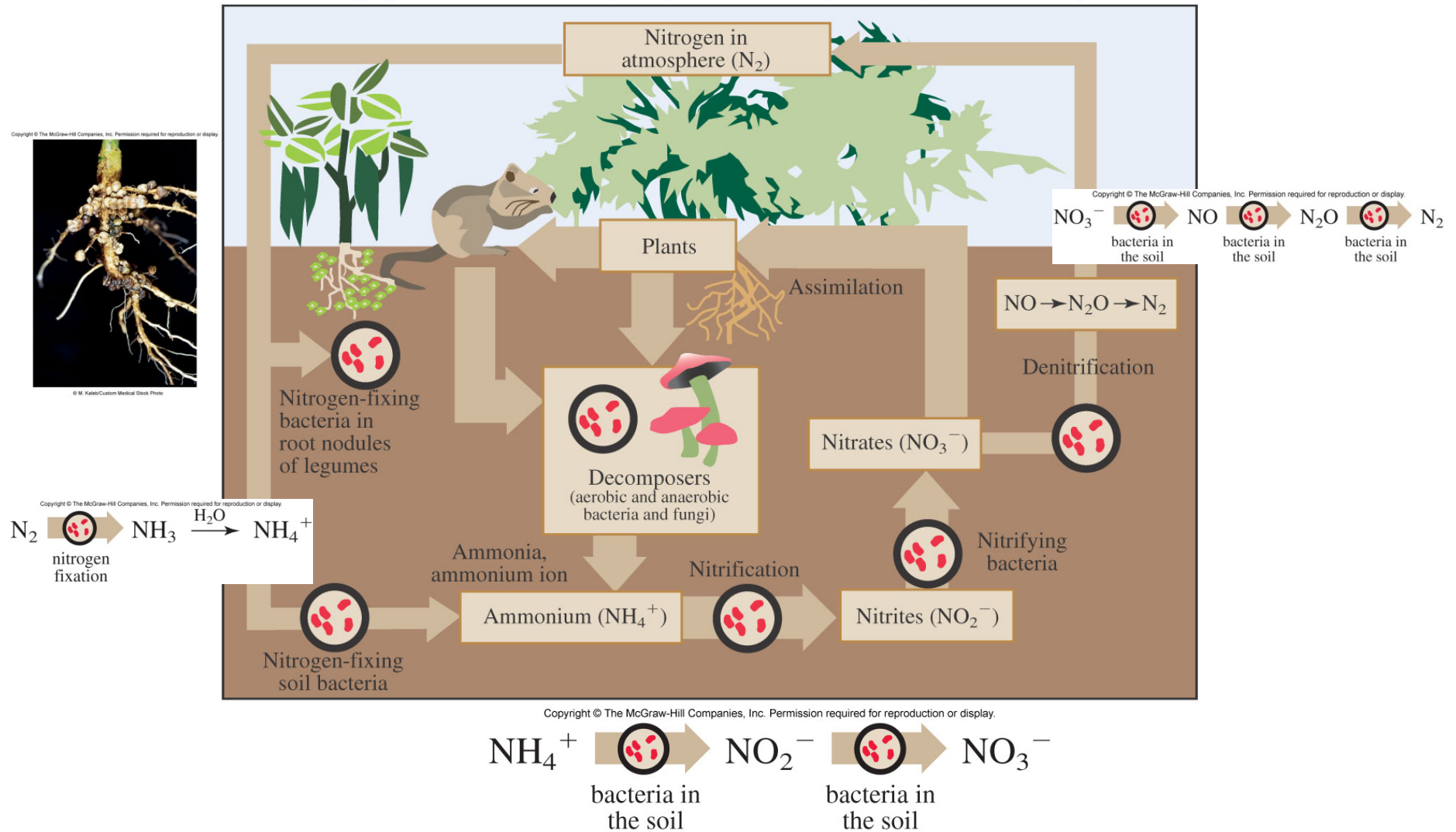


- 3) Denitrification



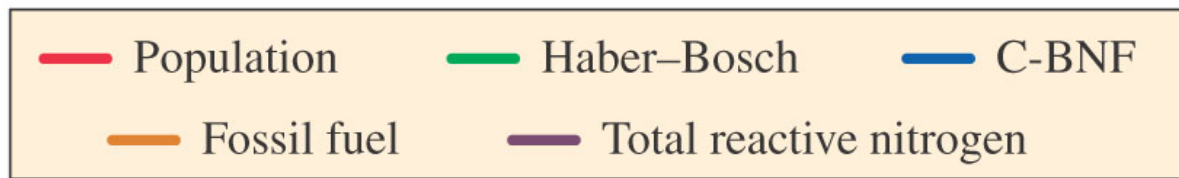
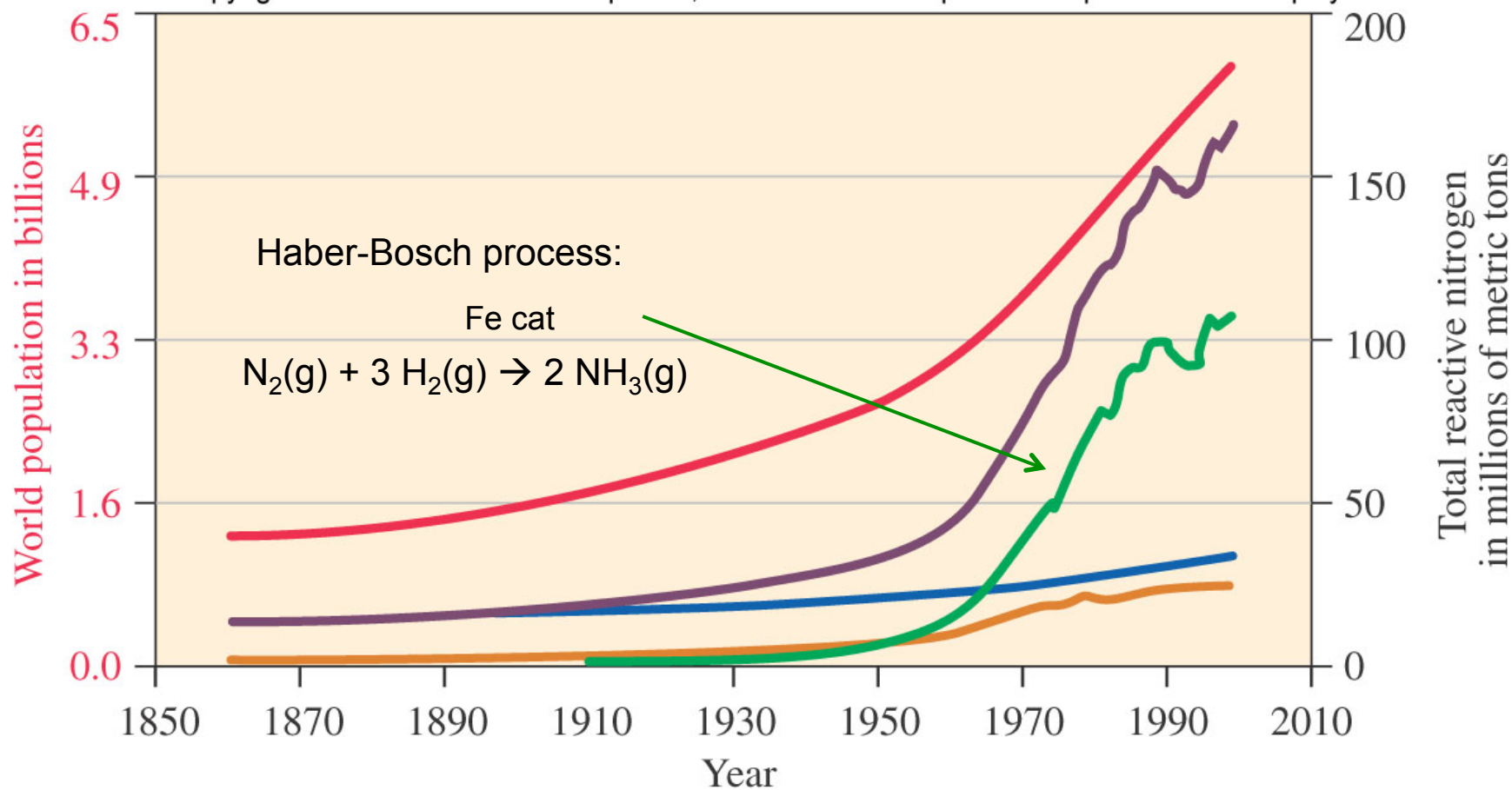
# The Nitrogen Cycle

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# Link Between Reactive Nitrogen and Population Growth

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C-BNF = reactive nitrogen created from cultivation of legumes, rice, and sugarcane